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A U.S. Department of Energy laboratory managed by The University of Chicago New homologous series of oxygen vacancy-ordered perovskite manganites: Sr_NMn_NO_{3N-2}

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Overview

Background, vacancy-ordered perovskite phases in the SrMnO_x system

- Oxygen-vacancy, charge and orbital ordering in $Sr_2Mn_2O_5$, $Sr_5Mn_5O_{13}$ and $Sr_7Mn_7O_{19}$
- A new homologous series $Sr_N Mn_N O_{3N-2}$
- Other systems showing $A_N B_N O_{3N-2}$ -type ordering
- Conclusions



Background

- $\blacksquare Sr_2Mn_2O_5$
 - 1985: Preparation and structural refinement (X-rays) Mn³⁺ in square-pyramidal coordination (Ca₂Mn₂O₅ type phase)
 - 1997: Magnetic and nuclear structure from neutron data.
 Orbital ordering in Mn³⁺ cations also proposed.

$\blacksquare Sr_5Mn_5O_{13}$

- 2003: conference report on phase $Sr_5Mn_5O_{13}$ with $BaLa_4Cu_5O_{13}$ tetragonal structure.
- Charge ordering of Mn³⁺/Mn⁴⁺ proposed, however, no paper followed

Caignaert V. et al, Mat. Res. Bull. 20 (1985) 479-484. V. Caignaert, J. Magn. Magn. Mater. 166 (1997) 117-123. Leligny et al (2003) Association Française de Cristallographie Meeting





V. Caignaert / Journal of Magnetism and M.



Background:

Structures of Sr₅Mn₅O₁₃ (SrMnO_{2.6}) and the new Sr₇Mn₇O₁₉ (SrMnO_{2.714}) phases were determined using synchrotron X-rays and neutron TOF data from multi-phase SrMnO₂ samples.





Background: Sr₅Mn₅O₁₃ and Sr₇Mn₇O₁₉







Sr Mn O

Charge ordering in $Sr_5Mn_5O_{13}$ and $Sr_7Mn_7O_{19}$





Orbital ordering in $Sr_5Mn_5O_{13}$ and $Sr_7Mn_7O_{19}$



- Chemical formula $Sr_2Mn_2O_5$ should be re-written $Sr_4Mn_4O_{10}$.
- Sr₄ Mn_4O_{10} , Sr₅ Mn_5O_{13} and Sr₇ Mn_7O_{19} comply with the notation Sr_N Mn_NO_{3N-2} with N=4, 5, and 7, respectively, for structures formed by 4 Mn³⁺ pyramids and N-4 Mn⁴⁺ octahedra.
- Is this a new homologous series? What are building rules?
- Phases in the SrMnO_x series should have oxygen contents x related to N by x=3-2/N: N=4 ⇒ x=2.5, N=5 ⇒ x=2.6, N=6 ⇒ x=2.667, N=7 ⇒ x=2.714, N=8 ⇒ x=2.75
- Is there other series of oxygen deficient perovskites (ABO_x) displaying the same structural features?



- Models of Sr₄Mn₄O₁₀ and Sr₅Mn₅O₁₃ can be built by joining N=4 and N=5 building blocks formed by 4 pyramids and N-4 octahedra with 2/m symmetry.
- Building blocks allow for charge and orbital ordering associated with the oxygen vacancy ordering.







Two different blocks of 4 pyramids and N-4=2 octahedra with 2/m symmetry can generate the hypothetical compounds $Sr_6Mn_6O_{16}$.



Two possible structures associated with N=6 blocks, not observed



Two blocks of 4 pyramids and N-4=3 octahedra with 2/m symmetry can generate compound $Sr_7Mn_7O_{19}$.





Only one structural pattern is formed from both blocks



For N>7 three or more blocks lead to two or more $Sr_N Mn_N O_{3N-2}$ compounds





Other systems showing $A_N B_N O_{3N-2}$ -type ordering:

LaCuOx & NdCuOx systems



- N=4 observed for d⁹ Cu²⁺ (pyramids)
- N=5 observed for d⁹ Cu²⁺ and d⁸ Cu³⁺ (octahedra). Large monoclinic distortion observed
- No phases with N>5 observed

Bringley et al, *Letters to Nature* (1990) 347, 263-265 Chen et al, *Inorg Chem.* (1995) 34, 2077-2083.

CaMnOx system

■ *Ca₂Mn₂O₅* (N=4) structure has been determined

■HREM and ED studies of CaMnO_{2.667} and CaMnO_{2.75} have shown formation of local structures with unit cells compatible with those proposed for N=6 ($Ca_6Mn_6O_{16}$) and N=8 ($Ca_8Mn_8O_{22}$) members of the series respectively

A neutron powder diffraction of CaMnO_{2.75} was inconclusive possibly due to the coexistence of multiple ordering arrangements

Poeppelmeier et al *J. Solid State Chem.* (1982) 45, 79-79. Reller et al *Proc. R. Soc. Lond. A* (1984) 349, 223-241. Chiang & Poeppelmeier, *Mater Lett.* (1991) 12, 102-108.

Size and charge of A-site cation appears to play a key role in the stabilization of different structural patterns.



Summary of observed phases $A_N B_N O_{3N-2}$

Blocks formed by 4 pyramids and N-4 octahedra with 2/m symmetry give rise to a series of compounds that could display orbital and charge ordering associated with the oxygen vacancy-ordering.

Building blocks	Ν	Cell setting, S.G.	Latt. param. (a, b, γ≠90)	Observed
	4	Orthorhombic, Pbam	$\sqrt{2}a_{\mathrm{P}}, 2\sqrt{2}a_{\mathrm{P}}$	(Sr/Ca)/Mn/O10 (La/Nd/Sr)/Cu/O10
	5	Tetragonal, P4/m	$\sqrt{5}a_{\rm P}, \sqrt{5}a_{\rm P}$	Sr ^o Mn ^o O13 La/Nd)©Cu©O13 (P2/m
	6	Orthorhombic, Pbmm	$\sqrt{2}a_{\rm P}, 3\sqrt{2}a_{\rm P}$	CaMnO2.667 ?
		Monoclinc, P2/m	$2a_{\rm P}, \sqrt{10}a_{\rm P}, 108.4^{\circ}$	
	7	Monoclinic, P2/m	$\sqrt{5}a_{\rm P}, \sqrt{10}a_{\rm P}, 98.13^{\circ}$	Sr7Mn7O19
	8	Monoclinc, P2/m	$2\sqrt{2}a_{\rm P},\sqrt{10}a_{\rm P},116.6^{\circ}$	
		Orthorhombic, Pbam	$\sqrt{2}a_{\mathrm{P}}, 4\sqrt{2}a_{\mathrm{P}}$	CaMnO2.75?



Conclusions & Outlook

New homologous series Sr_NMn_NO_{3N-2} where Jahn-Teller orbital-ordered d⁴ Mn³⁺ pyramids coexist with d³ Mn⁴⁺ octahedra has been described by simple model of building blocks formed by 4 pyramids and N-4 octahedra.

 \Rightarrow Single-phase samples are under preparation to study magnetic and transport properties.

■ *N*=4 and possibly *N*=6 and 8 phases observed in $Ca_N Mn_N O_{3N-2}$ system.

 \Rightarrow Extension of the model to include asymmetric building blocks may explain multiphase $_{\rm x}$

behavior of CaMnO system.

N=4 and N=5 phases observed in the LaCuO_x and NdCuO_x systems show analogous charge and orbital order of d⁹ Cu²⁺ pyramids and d⁸ Cu³⁺ octahedra.

⇒ Better understanding of electronic configuration relations among *d* and *d* manganates and cuprates could possibly be established.

A-site cationic charge and size play a key role in the stabilization and distortion of variable members of the series.
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Collaborators

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 - Stan Kolesnik (magnetism)

At IPNS

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- At APS
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 - Álvaro W. Mombrú & Ricardo Faccio (modelling)

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In memoriam



James D. Jorgensen, 1948 - 2006

